A CONVENIENT SYNTHESIS OF UNSYMMETRICAL BIBENZYLS, HOMOALLYLARENES, AND HOMOPROPARGYLARENES VIA PALLADIUM-CATALYZED CROSS COUPLING¹

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SUMMARY The palladium-catalyzed homobenzyl-aryl, homobenzyl-alkenyl, homoallyl-aryl, and homopropargyl-aryl cross coupling reactions provide the desired coupling products in high yields, whereas similar reactions involving alkynyl halides lead to low product yields

Organic compounds having two carbon-carbon double and/or triple bonds in a 1,5-relationship (1), such as 1,5-dienes and homoallylarenes, represent a wide variety of natural products. Until recently, construction of such units via cross coupling had been achieved mostly by joining two allylic, propargylic, and/or benzylic groups.² Although satisfactory in many cases, this approach has nevertheless been limited by difficulties such as cross-homo scrambling as well as regio- and/or stereoisomerization.³

$$\begin{pmatrix} Ar - \\ R^{a}R^{b}C = CR^{c} - \\ RC = C - \end{pmatrix} \xrightarrow{CH_{2}CH_{2}} \begin{pmatrix} -Ar' \\ -R^{d}C = CR^{e}R^{f} \\ -C = CR' \end{pmatrix}$$

We recently reported a Pd-catalyzed cross-coupling reaction of homoallyl- or homopropargylzincs with alkenyl halides as an efficient and selective route to 1,5-dienes and 1,5-enynes.⁴ We now report that the scope of this methodology may be extended to the homobenzyl-aryl (eq 1), homobenzyl-alkenyl (eq 2), homoallyl-aryl (eq 3), and homopropargyl-aryl (eq 4) cases, where the first entry in each case represents the corresponding organozinc reagent

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Although we previously noted that the use of Ni(PPh₃)₄ as a catalyst in conjunction with acetylenic reagents led to very low product yields,⁴ both Ni(PPh₃)₄ and Pd(PPh₃)₄ appear to give comparable results in other instances Except for a minor formation (<5-10%) of the dimers of aryl or alkenyl halides, these reactions are essentially free of other side reactions, such as β elimination Unfortunately, our attempts to use alkynyl halides have resulted in disappointing results. Thus, for example, the reaction of phenylethylzinc chloride with 1-octynyl bromide or iodide in the presence of 5 mol % of Pd(PPh₃)₄ produced only 30-40% yields of 1-phenyl-3-decyne contaminated with significant amounts of phenylethyl halides and 1-octyne, which presumably formed via metal-halogen exchange Use of Cl₂Pd(Ph₂P(CH₂)₃PPh₂)⁵ did not improve the product yield.

In sharp contrast with the favorable results described above, our attempts to synthesize compounds represented by 1 via formation of the carbon-carbon bond between the two methylene groups have led to mixtures of cross- and homo-coupled products. Among them are the Pd-catalyzed reactions of benzylzinc bromide with *a*-bromo-*p*-xylene, allyl bromide, and propargyl bromide

The following synthesis of (*E*)-1-phenyl-4-methyl-3-decene is representative To dry ZnCl (1 36 g, 10 mmol) and Mg (0 36 g, 15 mmol) covered with THF (10 mL) under nitrogen atmosphere was added 2-phenylethyl bromide (2.78 g, 15 mmol) in 10 mL of THF. The mixture was heated for 10 n at 50°C and cooled. To this was added sequentially $Pd(PPh_3)_4$ (0 404 g, 0.35 mmol) dissolved in 4 mL of THF and (*E*)-1-10do-2-methyl-1-octene The mixture was stirred for 3 h at 25°C and quenched with 3N HCl. Extraction (ether), washing (aq NaHCO₃), drying (MgSO₄), and distillation gave 1.48 g (92%) of (*E*)-1-phenyl-4-methyl-3-decene bp 101°C (0 3 mm), n²⁶D 1 4928, ¹H NMR (CDCl₃, Me₄S1) δ 0 89 (t, 7 Hz, 3 H), 1.1-1.4 (m, 8 H), 1 53 (s, 3 H), 1.96 (t, 7 Hz, 2 H), 2.64 (t, 7 Hz), 5.18 (t, 7 Hz, 1 H), 7.22 (m, 5 H) ppm.⁶

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- 5. It has been reported that the use of Cl₂Pd(Ph₂P(CH₂)₃PPh₂) and other Pd-phosphine complexes containing bidentate phosphine ligands leads to higher alkylation yields [Hayashi, T., Konishi, M, Kumada, M *Tetrahedron Lett.*, <u>1979</u>, 1871].
- 6. The ¹³C NMR spectrum of this compound is available (≥98% E) All products were adequately characterized by ¹H NMR, IR, and high resolution mass spectrometry For the determination of the stereochemistry of trisubstituted alkenes, see Rand, C L, Van Horn, D E, Moore, M W, Negishi, E J. Org. Chem 1981, 46, 4093. (Received in USA 5 May 1983)